New route synthesis of styryl pyrones:High yield synthesis, reactions and spectral properties of 2-phenyl-6-styryl-4-pyrones Abdel Moneim El-Ghanam

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2-Phenyl-6-styryl-4-pyrone derivatives have been synthesised from the reaction of ethyl phenylpropiolate with benzylideneacetone derivatives in the presence of sodium ethoxide. Treatment of styrylpyrone with phosphorus pentasulfide gave the corresponding styrylpyran-4-thione which on treatment with hydroxylamine hydrochloride and aqueous methylamine afforded the corresponding oxime and 1-methyl-2-phenyl-6-styrylpyridine-4(1*H*)-thione, respectively. On the other hand, styrylpyran-4-thione reacted with malononitrile to give pyrolylidenemalono-nitrile which on treatment with bidentate reagents, hydrazine hydrate, hydroxylamine hydrochloride, thiourea, and thiosemicarbazide afforded the corresponding styrylspiropyran derivatives of pyrazole, 1,2-oxazole or 1,3-thiazine, respectively.

Keywords: styrylpyrones, spectral properties

Many 4-pyrones or compounds containing 4-pyrone moieties has been synthesised during the last several decades, which have biological activities, such as herbicidal, fungicidal, antiallergenic as well as anticancer activity.¹⁻³ The 4-pyrones containing styryl groups attached to the pyrone ring has been reported to have anticancer activity.^{4,5} Also, 2-hydroxy-5-styryl-4-pyrone has been used in the formulation of skin-lightening cosmetic⁶ as well as in the synthesis of polycondensed heterocyclic compounds.⁷ Several 2-hydroxy-5-arylethenyl-4pyrones as well as 5-methoxy analogues have been prepared by condensation of benzaldehyde with allomaltol monomethyl ether⁸ or by Wittig reaction of 2-chloromethyl-5-methoxy-4pyrone with triphenylphosphine⁹ followed by the reaction with aryl aldehydes, respectively. The yields of these reactions did not exceed 36%. The purpose of the present work was to find a new synthetic procedure to prepare and optimised the yields of a series of 6-styryl-4-pyrones due to their importance in biological activity, specially as anticancer activities.

Results and discussion

The titled compound has been synthesised by stirring a mixture of ethyl phenylpropiolate **1** and benzylideneacetone derivatives **2a–d** in dry ether at 0° C in 1:1 molar ratio using sodium ethoxide as a base for 5 h. The reaction mixture was



Ar: $a = C_6H_5$, $b = p-CH_3C_6H_4$, $c = p-CH_3OC_6H_4$, $d = p-CIC_6H_4$

Scheme 1

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poured in ice water and acidified with concentrated HCl to give 2-phenyl-6-styryl-4*H*-pyran-4-ones **4a–d** (Scheme 1). The products, 2-phenyl-6-styryl-4-pyrones, probably formed through the intermediates **3**.¹⁰ The structure of these styryl derivatives **4a–d** were elucidated from their spectral and analytical data (see Experimental). The IR spectra showed bands for the carbonyl of 4-pyrones at 1668–1638 cm⁻¹ and a band at 1615–1592 cm⁻¹ characteristic to a double bonds of side chain styryl groups. ¹H NMR spectra, showed beside other characteristics two doublets at δ 6.49–6.92 and δ 6.84–7.62 with the coupling constants 16.2–16.7 Hz proving the *trans*-configuration for the ethylenic double bonds of the styryl moiety at position 2.⁸

Treatment of **4a** with phosphorus pentasulfide afforded 2-phenyl-6-styryl-4*H*-pyran-4-thione **5** which on treatment with hydroxylamine hydrochloride and aqueous methylamine afforded the oxime **6** and 1-methyl-2-phenyl-6-styrylpyridine-4(1*H*)-thione **7**, respectively (Scheme 2). While the treatment of thione **5** with malononitrile in the presence of triethylamine gave (2-phenyl-6-styryl-4*H*-pyran-4-ylidene)malononitrile **8** in moderate yield (Scheme 2). The ¹H NMR spectrum of the thione **5** showed an downfield shift in the resonance of the H-3 and H-5 protons, δ 7.92 and δ 8.36, respectively comparing to the parent styryl pyran **4a**. Similar deshielding on replacement of carbonyl with thiocarbonyl were also observed for other 4*H*-pyran-4-thiones.¹¹ Such significant deshielding in the resonance of H-3 and H-5 protons of **5** can be attributed to the increased magnetic anisotropy of the thione over carbonyl.¹²

The structure of 4-pyrolylidenemalononitrile **8** is confirmed from its spectral and analytical data (Experimental). The IR spectrum showed a moderately C=N absorption at 2188 cm⁻¹ while its ¹H NMR spectrum showed singlets at δ 6.82 and δ 7.22 for H-3 and H-5 protons of pyran ring besides two doublets at δ 7.06 and δ 7.26 for styryl moiety.

Treatment of **8** with hydrazine hydrate, hydroxylamine hydrochloride, thiourea and thiosemecarbazide in absolute ethanol in the presence of pipredine afforded the corresponding styryl spiropyran derivatives of pyrazole, isoxazole, 1,3-thiazine, **9–12**, respectively (Scheme 2). The reaction involve the addition of the amino or mercapto groups at the ethylenic double bond and nucleophilic attack of the amino group to give the cyclised spiro compounds.^{13,14} All styryl spiropyran derivatives **9–12** are confirmed from their spectral and analytical data (Experimental).

Experimental

Elemental analysis were preformed on a Perkin-Elmer 240 microanalyser. Melting points were recorded on a Kofler Block and are uncorrected. IR spectra were measured with a Unicam SP 1025 spectrophotometer for KBr pellets. The ¹H NMR spectra were recorded on Jeol ECA 500 MHz spectrometer using TMS as internal standard. Mass spectra were recorded on a Varian Mat CH-7 instrument at 70 eV.

Preparation of 2-phenyl-6-styryl-4H-pyran-4-ones **4a–d**. General procedure

An ethereal solution of benzylidene acetone derivatives 2a-d (0.0842 mol) and ethyl phenylpropiolate 1 (0.0842 mol) were added dropwise



Scheme 2

to an ice cold suspension of sodium ethoxide (0.0286 mol) in dry ether (120 ml). The reaction mixture was stirred at 08°C for 5 h and then poured to an ice cold water and then acidified with concentrated HCl to give 2-phenyl-6-styryl-4*H*-pyran-4-ones **4a–d** as yellow solid which crystallised from methanol.

2-Phenyl-6-styryl-4H-pyran-4-one (**4a**): Yield, 83 %, m.p. 146– 148 °C; IR (KBr, cm⁻¹): 3096, 1652, 1588; ¹H NMR (DMSO-d₆, δ): 6.48 (s, 1H, C-3 of pyrone), 6.66 (s, 1H, C-5 of pyrone), 7.08 (d,1H, J = 16.2 Hz, styrene proton), 7.42 (d, 1H, J = 16.2 Hz, styrene proton), 7.36–7.72 (m, 10H, aromatic); MS: m/z (relative abundance): M⁺ 274 (24), 258 (36), 246 (51) 197 (42), 143 (100), 120 (26), 94 (32), 77 (46). Anal. Calcd. For C₁₉H₁₄O₂: C, 83.21; H, 5.11. Found: C, 83.42; H, 5.02.

2-Phenyl-6-[2-(4-methylphenyl)ethenyl]-4H-pyran-4-one (4b): Yield, 79 %, m.p. 166–167 °C; IR (KBr, cm⁻¹): 3112, 2896, 1648, 1592; ¹H NMR (DMSO-d₆, δ): 2.44 (s, 3H, CH₃), 6.42 (s, 1H, C-3 of pyrone), 6.82 (s, 1H, C-5 of pyrone), 7.16 (d, 1H, J = 16.5 Hz, styrene proton), 7.62 (d, 1H, J = 16.5 Hz, styrene proton), 7.62 (d, 1H, J = 16.5 Hz, styrene proton), 7.26–7.64 (m, 9H, aromatic); MS: m/z (relative abundance): M⁺ 288 (28), 273 (12), 260 (38), 211 (28), 143 (100), 94 (36), 77 (42), 91 (22). Anal. Calcd. For C₂₀H₁₆O₂: C, 83.33; H, 5.56. Found: C, 83.42; H, 5.38.

2-*Phenyl-6-[2-(4-methoxyphenyl)ethenyl]-4H-pyran-4-one* (4c): Yield, 88 %, m.p. 196–198 °C; IR (KBr, cm⁻¹): 3118, 2904, 1638, 1615; ¹H NMR (DMSO-d₆, δ): 3.72 (s, 3H, OCH₃), 6.46 (s, 1H, C-3 of pyrone), 6.68 (s, 1H, C-5 of pyrone), 7.14 (d, 1H, *J* = 16.4 Hz, styrene proton), 7.54 (d, 1H, *J* = 16.4 Hz, styrene proton), 7.42–7.86 (m, 9H, aromatic); MS: *m/z* (relative abundance): M⁺ 304 (28), 273 (32), 143 (100), 107 (52), 94 (62), 77 (66). Anal. Calc. For C₂₀H₁₆O₃: C, 78.95; H, 5.26. Found: C, 78.82; H, 5.30.

2-Phenyl-6-[2-(4-chlorophenyl)ethenyl]-4H-pyran-4-one (4d): Yield, 82 %, m.p. 170–172 °C; IR(KBr, cm⁻¹) 3102, 2896, 1668, 1615; ¹H NMR (DMSO-d₆, δ): 6.38 (s, 1H, H-3 of pyrone), 6.64 (s, 1H, H-5 of pyrone), 7.12 (d, 1H, J = 16.7 Hz, styrene proton), 7.38–7.82 (m, 10H, 1H of styrene + 9H aromatic). MS: m/z (relative abundance): M⁺ 310 (6), 308 (19), 280 (36), 231 (26), 143 (100), 112 (42), 94 (48), 77 (28). Anal. Calc. For C₁₉H₁₃O₂Cl: C, 73.91; H, 4.21. Found: C, 73.96; H, 4.12.

2-Phenyl-6-styryl-4H-pyran-4-thione (5): This compound was prepared from 2-phenyl-6-styryl-4H-pyran-4-one 4a and phosphorus pentasulfide as described earlier.¹²

Yield, 71%, m.p. 182–183 °C; IR (KBr, cm⁻¹): 3042, 1648, 1082; ¹H NMR (CDCl₃, δ): 8.36, 7.92 (s, 2H, H-3 and H-5 of pyran), 7.12 (d, 1H, *J* = 16.2 Hz), 7.38–7.78 (m, 11H, 10H aromatic + 1H styrene); MS: *m*/z (relative abundance): M⁺ 290 (20), 246 (30), 197 (36), 143 (100), 120 (18), 94 (46), 77 (52). Anal. Calc. For C₁₉H₁₄OS: C, 78.62; H, 4.83. Found: C, 78.54; H, 4.92.

2-Phenyl-6-styryl-4H-pyran-4-one oxime (6): This compound was prepared from 2-phenyl-6-styryl-4H-pyran-4-thione 5 and hydroxyl-amine hydrochloride and sodium acetate in ethanol as described earlier.¹²

Yield, 71%, m.p. 168 °C; IR (KBr, cm⁻¹): 3528 (br), 3086, 1652; ¹H NMR (CDCl₃, δ): 7.58 (m, 10H, aromatic), 7.14, 6.72 (s, 2H, C-3 and C-5 pyran protons) 7.02 (d, 1H, J = 16.2 Hz), 7.22 (d, 1H, J = 16.2 Hz). Anal. Calc. For C₁₉H₁₅NO₂: C, 78.89; H, 5.19; N, 4.84. Found: C, 78.71; H, 5.02; N, 4.96.

1-Methyl-2-phenyl-6-styrylpyridine-4(1H)-thione (7): This compound was prepared from styrylpyranthione **5** and 33% aqueous methylamine solution as described earlier.¹⁰

Yield, 73%, m.p. 156 °C; IR (KBr, cm⁻¹): 3088, 2976, 1634, 1076; ¹H NMR (CDCl₃, δ): 7.62 (m, 10H, aromatic), 7.08 (d, 1H, J = 16.2Hz), 7.22 (d, 1H, J = 16.2 Hz), 3.36 (s, 3H, CH₃). Anal. Calc. For C₂₀H₁₇NS: C, 79.21; H, 5.61; N, 4.62; S, 10.56. Found: C, 79.26; H, 5.72; N, 4.70; S, 10.38.

(2-Phenyl-6-styryl-4H-pyran-4-ylidene)malononitrile (8): 0.023 mol of 2-phenyl-6-styryl-4H-pyran-4-one 4a was added to an equimoler amount of malononitrile in 40 ml ethanol and few drops of trimethylamine. The reaction mixture was refluxed for 4 h, concentrated and cooled to give 8, which collected by filtration and crystallised from ethanol.

Yield, 71%, m.p. 166–168 °C; IR (KBr, cm⁻¹): 3082, 2188, 1646: ¹H NMR (DMSO-d₆, δ): 7.22, 6.82 (s, 2H, for H-3 and H-5 of pyran), 7.06 (d, 1H, *J* = 16.2 Hz), 7.26 (d, 1H, *J* = 16.2 Hz), 7.38–7.82 (m, 10H, aromatic). Anal. Calc. For C₂₂H₁₄N₂O: C, 81.99; H, 4.35; N, 8.70. Found: C, 81.82; H, 4.32; N, 8.82.

Synthesis of spiro compounds 9–12. General procedure

A solution of (2-phenyl-6-styryl-4*H*-pyran-4-ylidene)malononitrile **8** (0.023 mol) in 40 ml absolute ethanol was treated with equimolar amount of hydrazine hydrate, or hydroxylamine hydrochloride or thiourea or thiosemecarbazide and few drops of piperidine. The reaction mixture was refluxed for 4 hours, concentrated, cooled, and the separated compound was filtered off and recrystallised from ethanol.

5'-amino-2-phenyl-6-styryl-2',4'-dihydrospiro[pyran-4,3'pyrazole]-4'-carbonitrile (**9**): Yield, 62 %, m.p. 217–219 °C; IR (KBr, cm⁻¹): 3391, 3326, 3214, 3094, 2198, 1642; ¹H NMR (DMSO-d₆, δ): 10.82 (br, 1H, NH), 6.92 and 7.20 (s, 2H for H-3 and H-5 of pyran), 7.08 (d, 1H, *J* = 16.2 Hz), 7.28 (d, 1H, *J* = 16.2 Hz), 7.40–7.78 (m, 10H, aromatic), 5.42 (br, 2H, NH₂). Anal. Calc. For C₂₂H₁₈N₄O: C, 74.58; H, 5.08; N, 15.82. Found: C, 74.42; H, 5.16; N, 15.66.

5-Amino-2'-phenyl-6'-styrylspiro[[1,2]oxazole-3,4'-pyran]-4carbonitrile (**10**): Yield, 60 %, m.p. 196–198 °C; IR (KBr, cm⁻¹): 3398, 3322, 3196, 3092, 2186, 1642; ¹H NMR (DMSO-d₆, δ): 11.24 (br, 1H, NH), 7.62 (m, 10H, aromatic), 7.20, 6.88 (s, 2H, H-3 and H-5 of pyran), 7.12 (d, 1H, *J* = 16.2 Hz), 7.30 (d, 1H, *J* = 16.2 Hz), 5.62 (br, 2H, NH₂). Anal. Calc. For C₂₂H₁₇N₃O₂: C, 74.37; H, 4.79; N, 11.83. Found: C, 74.42; H, 4.86; N, 11.74.

2',4'-Diamino2-phenyl-6-styrylspiro[pyran-4,6'-[1,3]thiazine]-5'carbonitrile (11): Yield, 68 %, m.p. 182–183 °C; IR (KBr, cm⁻¹): 3390, 3336, 2202, 1638; ¹H NMR (DMSO-d₆, δ): 7.62 (m, 10H, aromatic), 7.22, 6.84 (s, 2H, C-3 and C-5 of pyran), 7.08 (d, 1H, J = 16.2 Hz), 7.14 (d, 1H, J = 16.2 Hz), 5.46–5.82 (br, 4H, 2NH₂). Anal. Calc. For C₂₃H₁₈N₄OS: C, 69.35; H, 4.52; N,14.07; S, 8.04. Found: C,69.28; H, 4.56; N, 14.12; S, 8.26.

4'-Amino-2'-hydrazino-2-phenyl-6-styrylspiro[pyran-4,6'-[1,3]thiazine]-5'-carbonitrile (12): Yield, 62 %, m.p. 242–244 °C; IR (KBr, cm⁻¹): 3408, 3332, 3236, 3092, 2936, 2196, 1636; ¹H NMR (DMSO-d₆, δ): 10.82 (br, 1H, NH), 7.68 (m, 10H, aromatic), 7.26, 6.88 (s, 2H, C-3 and C-4 of pyran), 7.06 (d, 1H, *J* = 16.2 Hz), 7.18 (d, 1H, *J* = 16.2 Hz), 4.88–5.46 (br, 4H, 2NH₂). Anal. Calc. For C₂₃H₁₉N₅OS: C, 66.83; H, 4.60; N, 16.95; S, 7.75. Found: C; 66.68; H, 4.62; N, 16.80; S, 7.82.

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